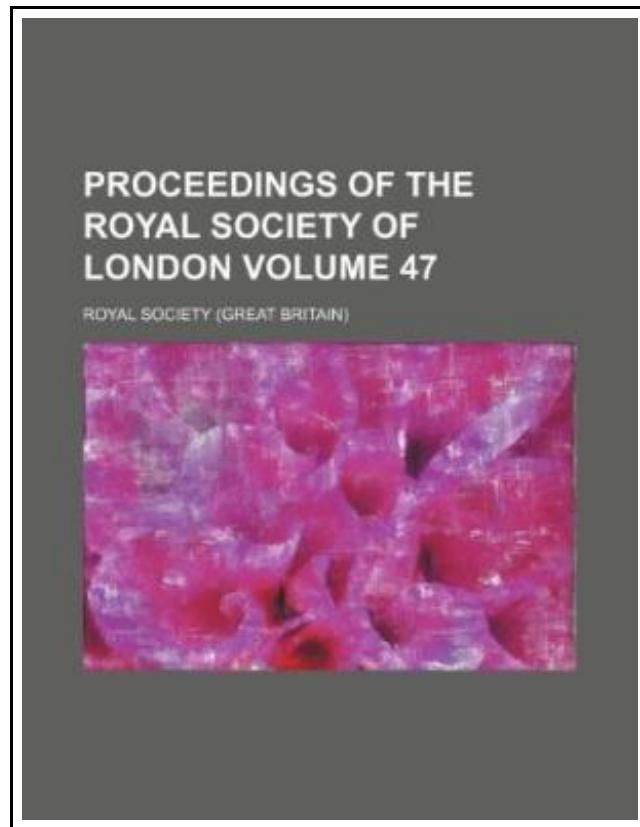


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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****.This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1890 Excerpt: .Nor need the thermal expansion of the substance any longer be considered. I describe the apparatus briefly here, as it seems in fact to eliminate to a great extent, if not entirely, what small risk of error obtains in the previously described apparatus. Its advantages, however, are most conspicuous in the use for which it was designed, the only use to which, up to the present I have applied it--the calorimetry of gases. Its application to this branch of physics must ever be its most important use. Indeed in the conditions obtaining in the calorimetry of solids or liquids it is hardly called for. However, even in these latter cases, where it is advisable to enclose the substance from contact with the steam, the differential calorimeter would enable us to effect the experiment somewhat more accurately than would be possible with the use of the single calorimeter. Plate 7 shows, in side sectional elevation, fig. 2, and front sectional elevation, fig. 1, to a scale of one-tenth, the differential calorimeter which I have at present in use in dealing with gases. The spheres, one of which is used to hold the gas, are shown hanging in the calorimeter. The drawing needs little explanation after what has been said about the single calorimeter. The principle is obvious. Apart from its special application to gases, it may be said that the calorimeter is so constructed that carriers depending from both arms of the balance are hung within it,...



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